



AALBORG UNIVERSITY
DENMARK

Utilization of organic resources and microbial electrochemistry for green hydrogen production

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Organic Resources & Microbial Pathways

1. Organic waste (household)

2. Wastewater

3. Produced water

4. Volatile fatty acids

1. Biofermentation

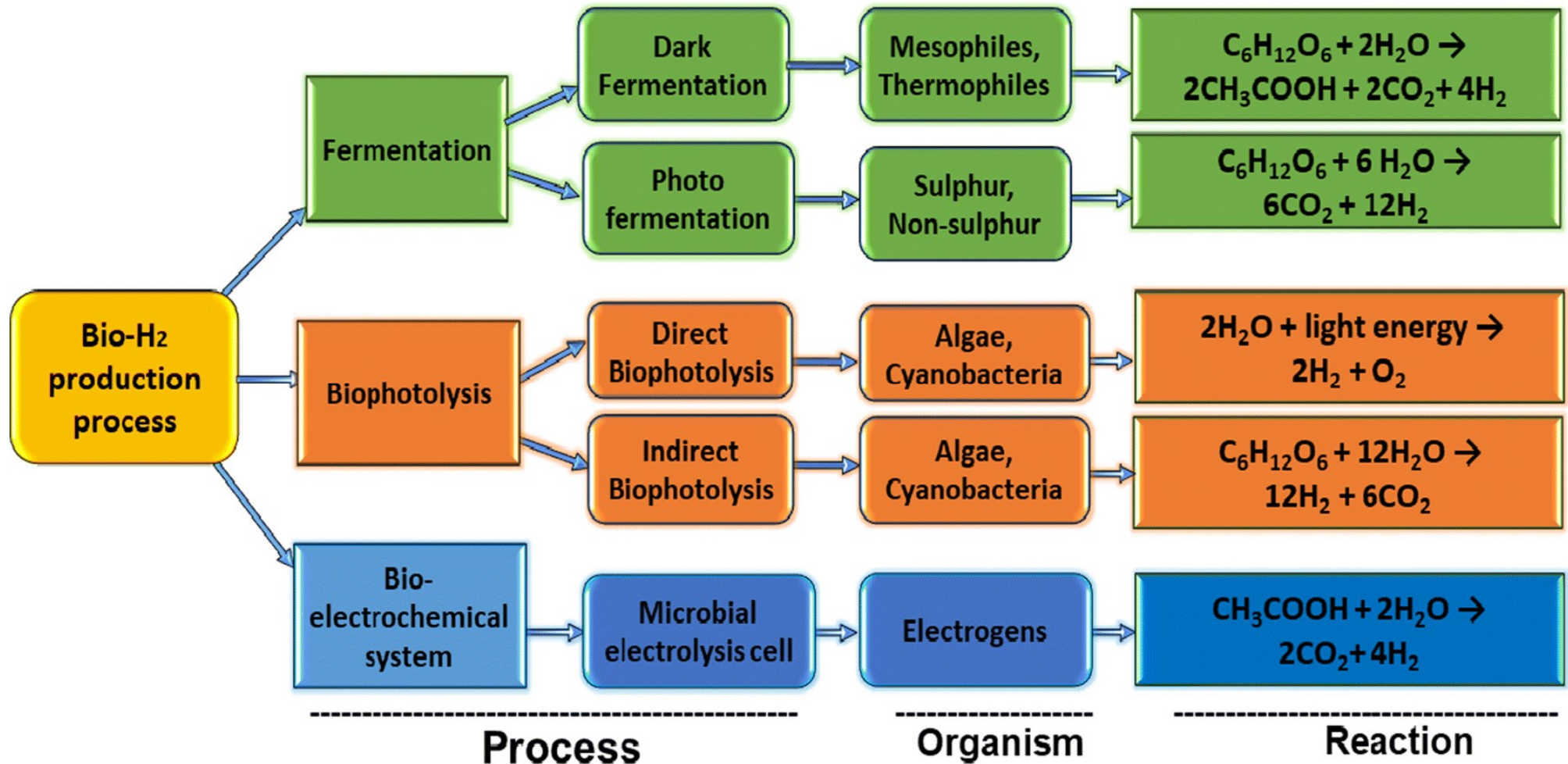
2. Biophotolysis

3. Bioelectrochemical system

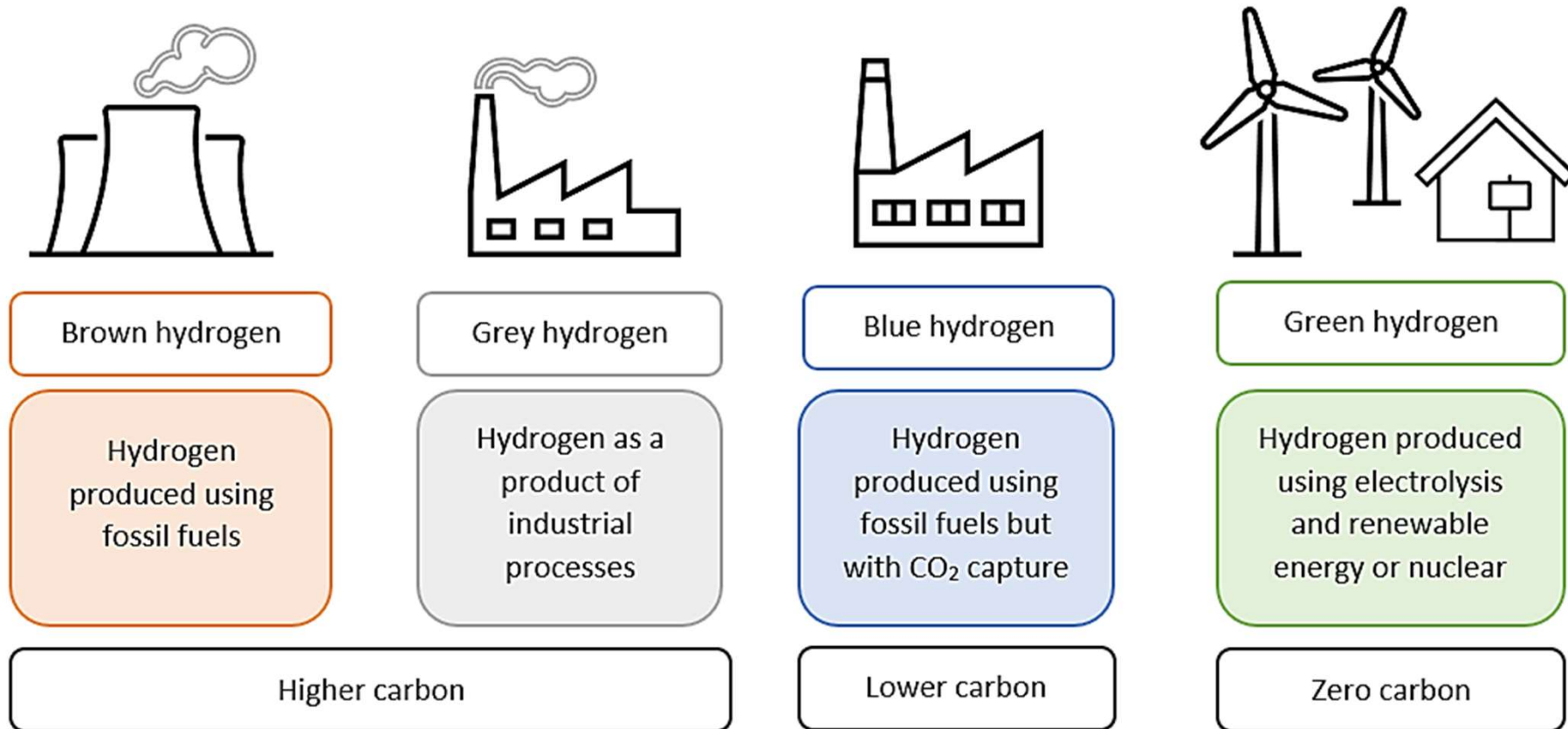
4. Gasification



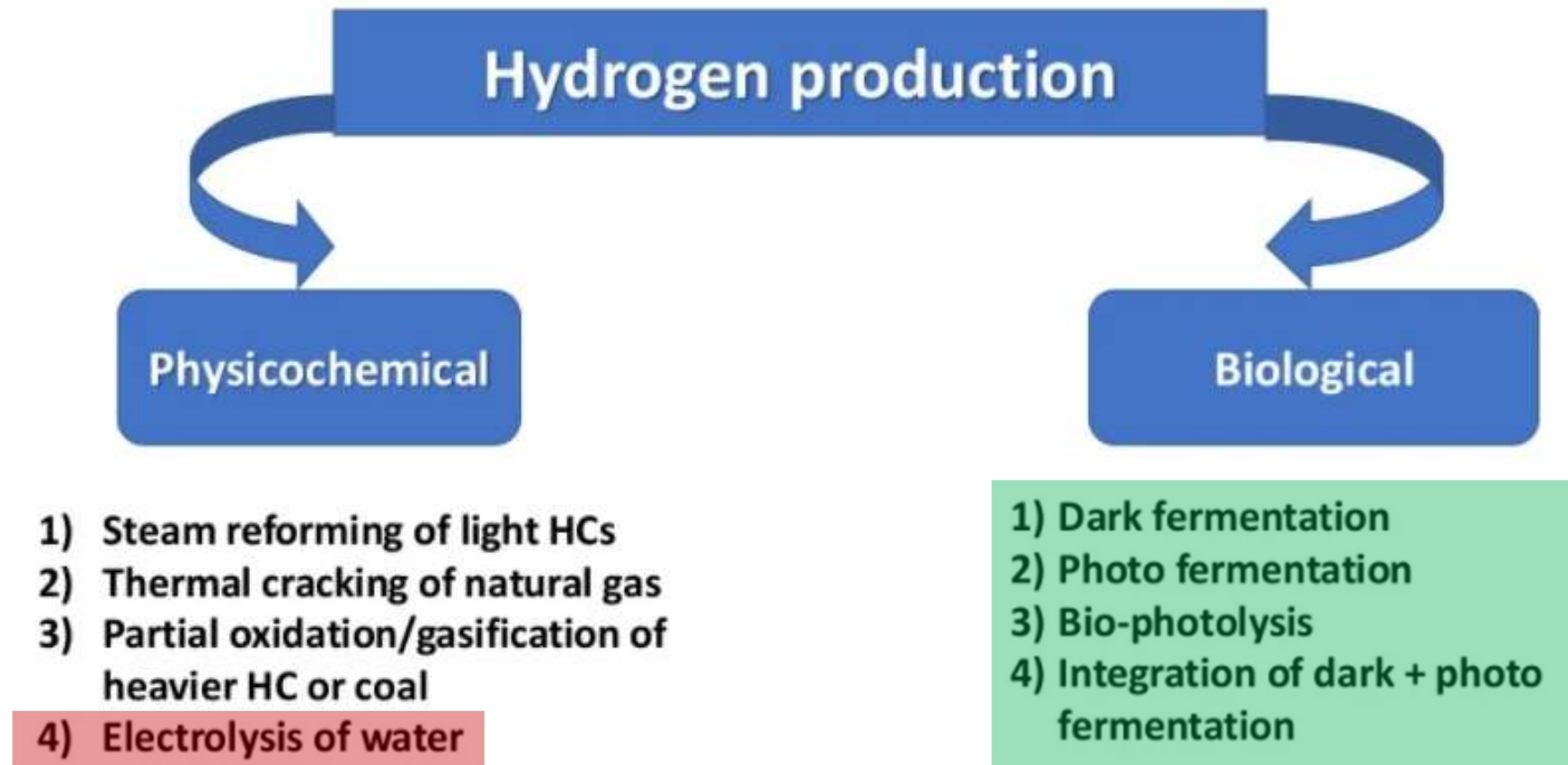
Microbial Pathways



‘Colors’ of Hydrogen



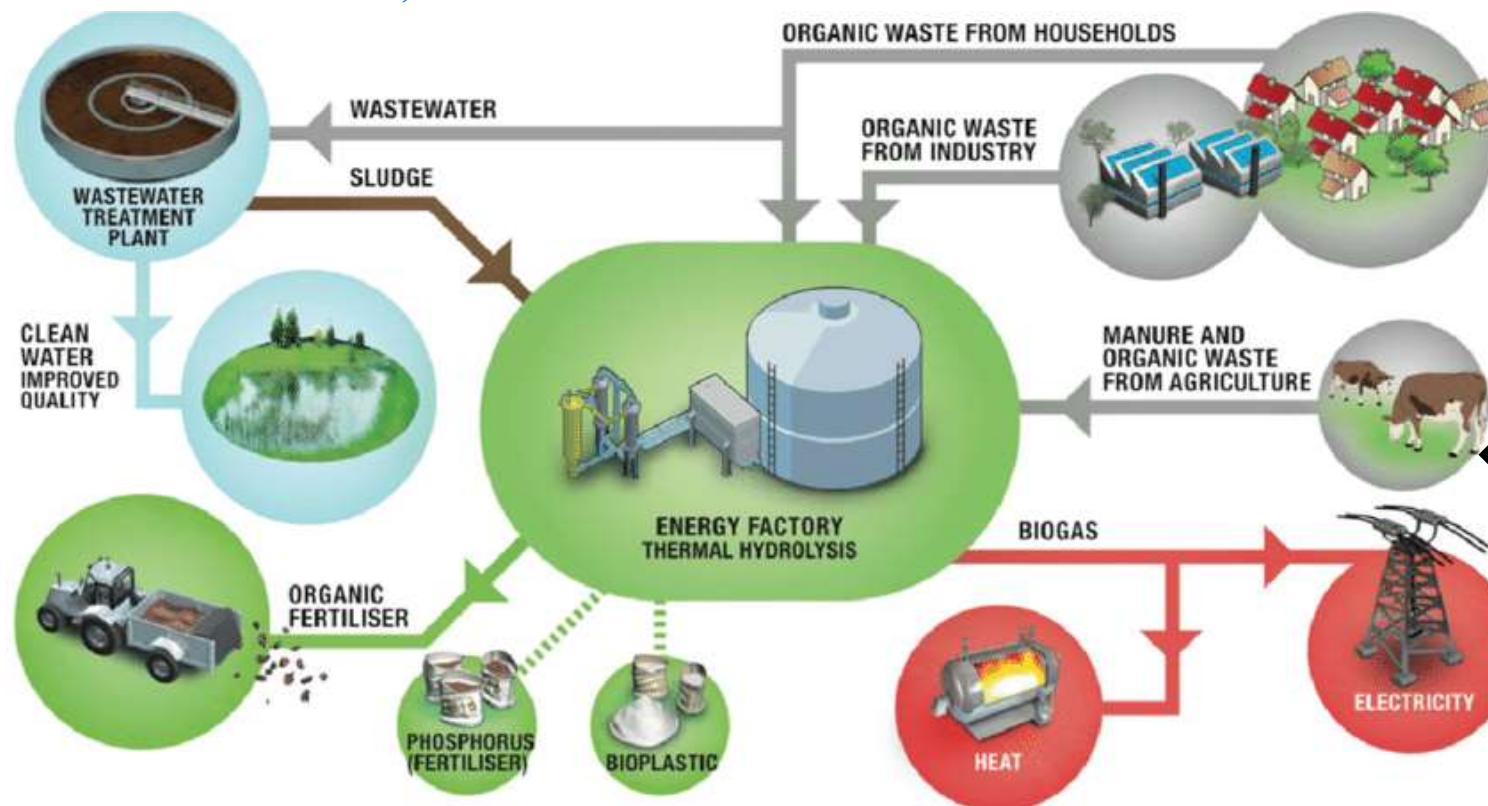
Biohydrogen Pathways



The Biorefinery Approach

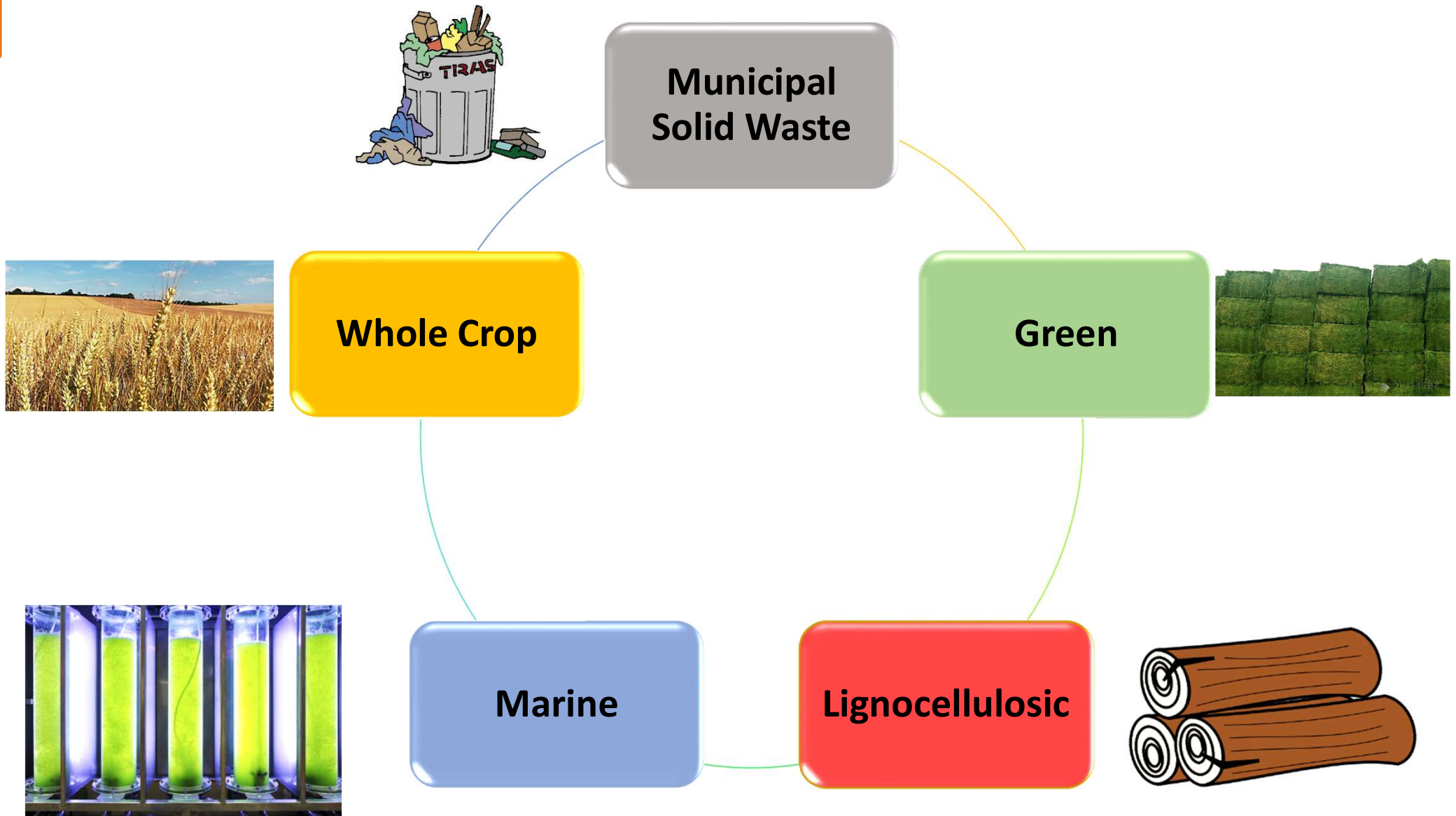
Biorefinery: Refinery that converts biomass to energy and other beneficial byproducts (such as chemicals).

Biorefining is defined as "the sustainable processing of **biomass** into a spectrum of **bio-based products** (food, feed, chemicals, materials) and **bioenergy** (biofuels, power and/or heat)"



**Billund
Biorefinery**

Type of Biorefineries





<https://flexigreenfuels.eu/>

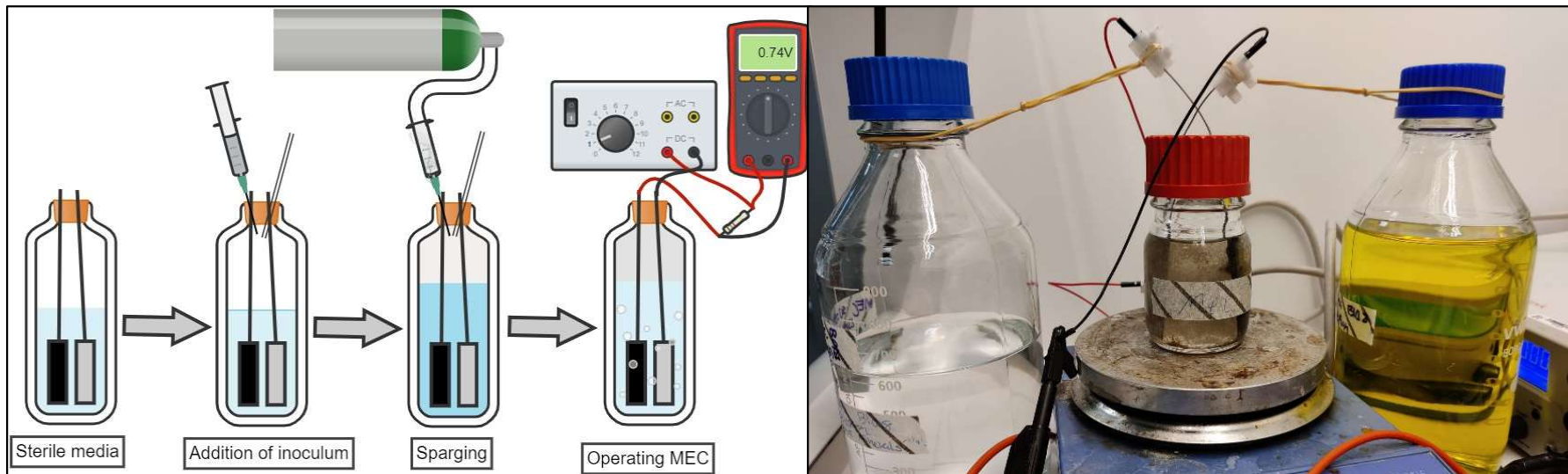
Project budget  ~4 Mio €	Project partners  13 Partners	Project Start  January 1st 2021	Project Duration  36 Months
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The Consortium

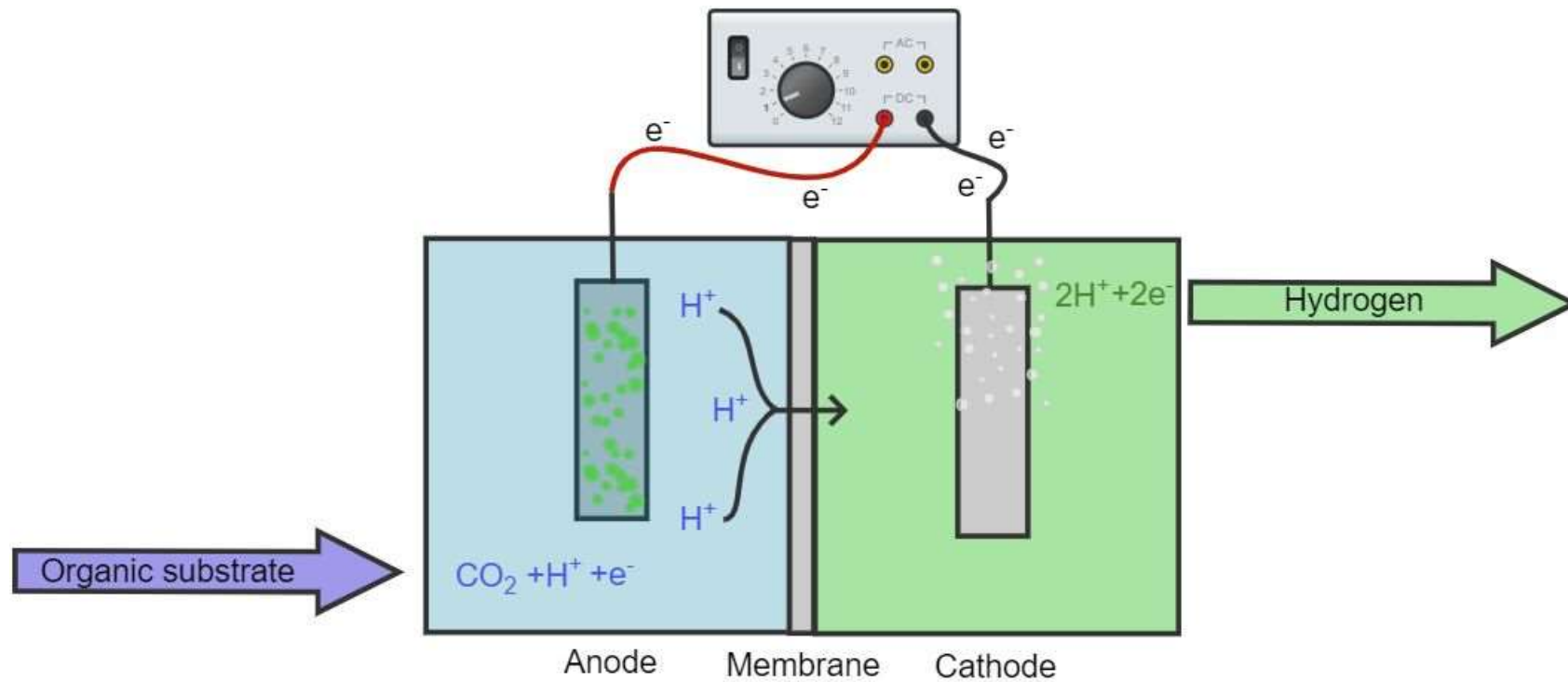


A Microbial Electrolysis Cell (MEC)

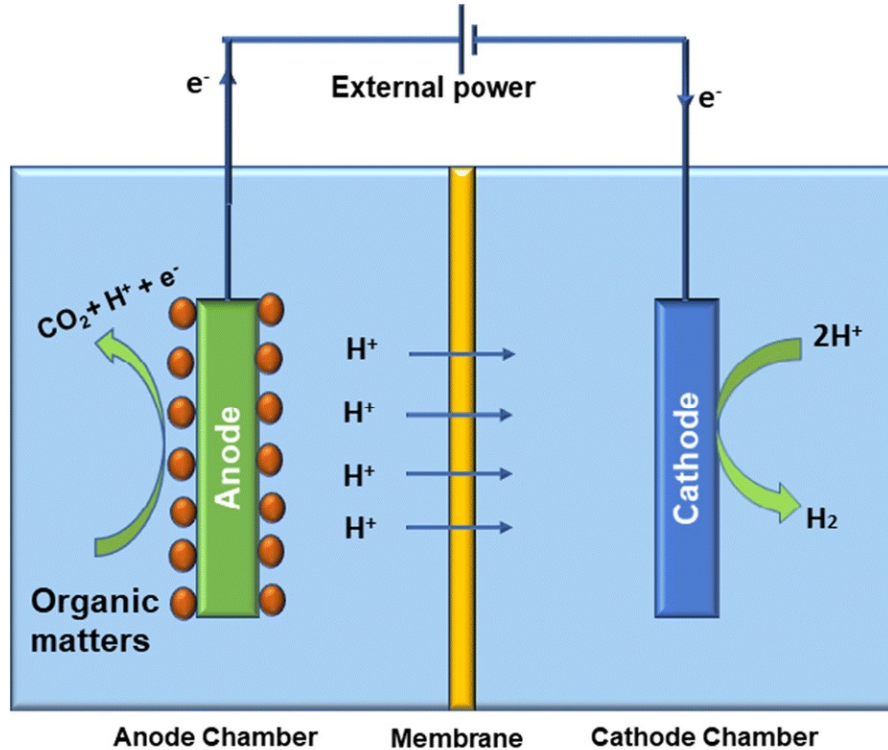
- Breakthrough in 2005
- Architecture:
 - Anode: A microbial anode
 - Cathode: An “almost conventional” hydrogen evolution
 - Electrolyte: microorganisms and the culture medium
 - Catalyst: electroactive biofilm i.e. electro-catalyst
- The microbial anode can oxidize a large variety of low-cost carbon compounds



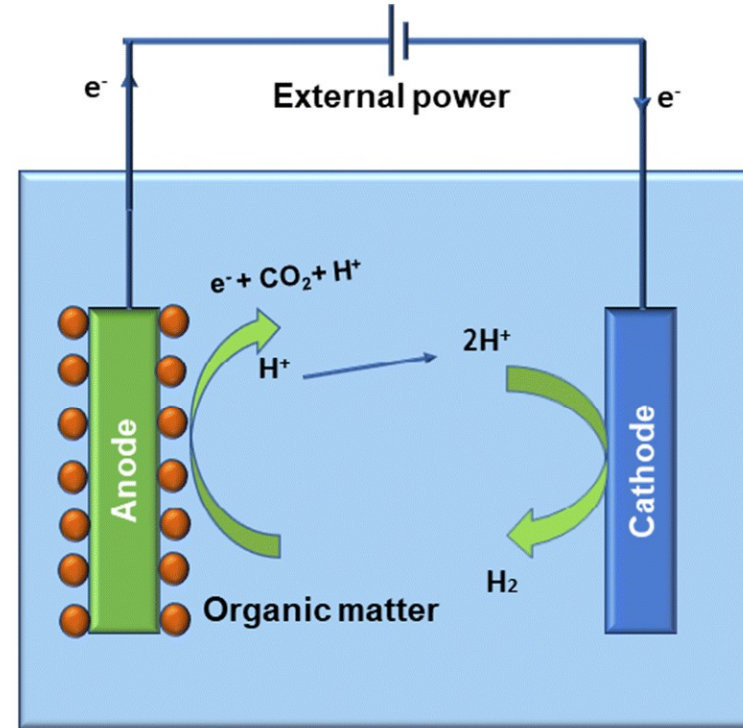
MEC Setup



MEC Design Configurations

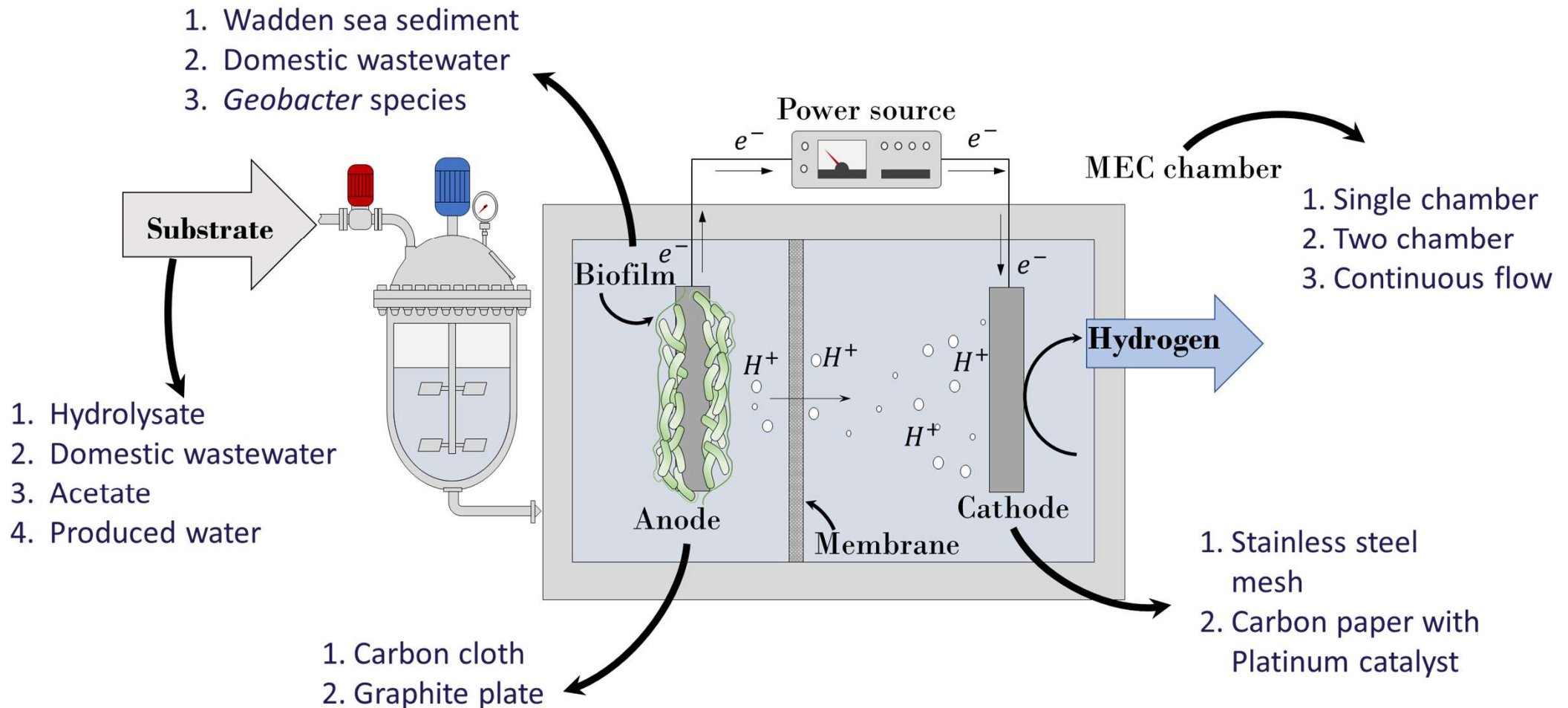


Two-chamber MEC



Single-chamber MEC

Various challenges in MEC research



Antagonistic Problems

Sustaining microbial growth in a culture medium

VS

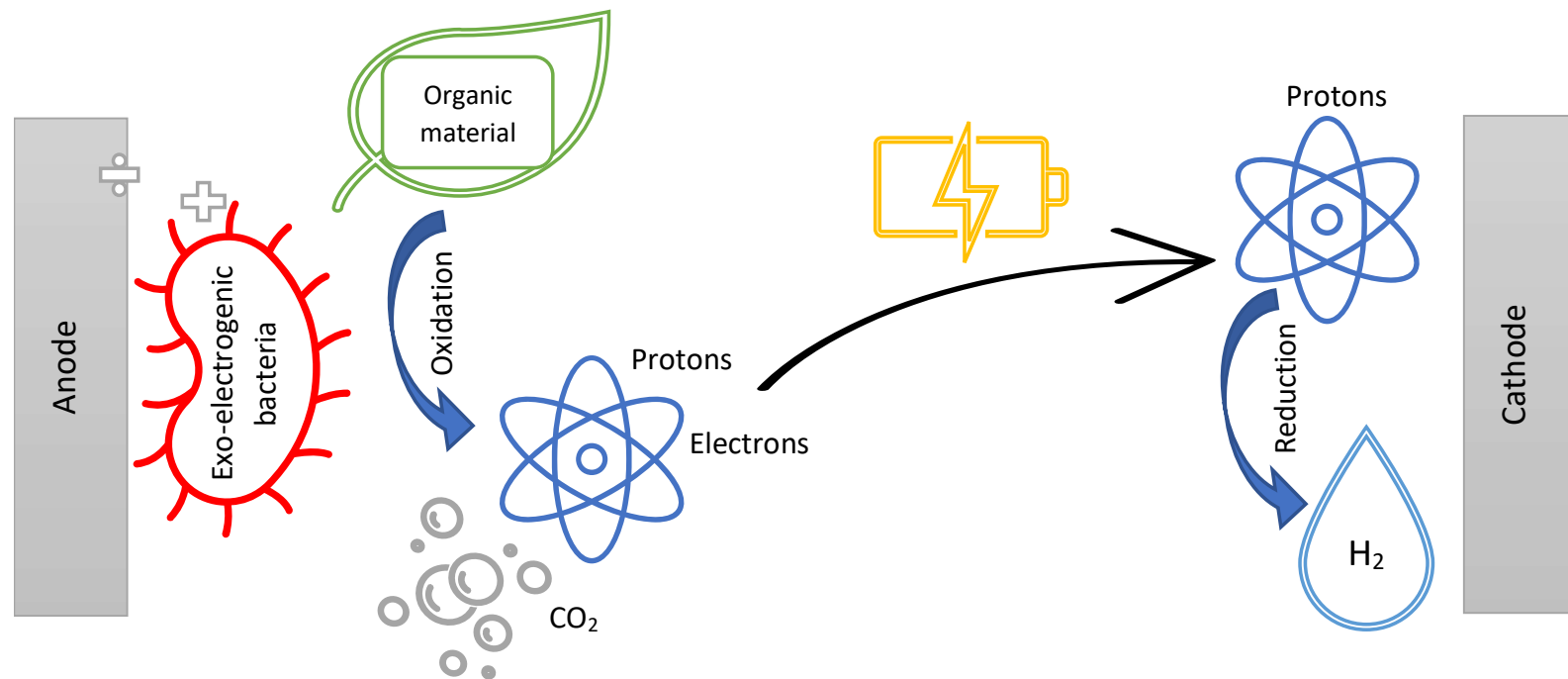
an electrolyser for large-scale hydrogen production are two antagonistic objectives.

- Electrochemical processes : Clean electrolytes, minimal inhibition of the electrode catalysts, high energy input.
- Sustaining microbial growth: Neutral pH, chemically rich media, slow start-up time .



Strengths of a MEC

- Low equilibrium cell voltage 0.123 V vs 1.23 V for water electrolysis
- Possibility to work at high pressure
- Anode: low-cost, self-assembled and self-sustained; heterogeneous organic electrolytes.
- Potential for methane (anaerobic digestion applications)



Areas of Improvement

- Hydrogen can be consumed at the bioanode
- Ca^{2+} , Mg^{2+} and other ions can cause local alkalization on the cathode
- Saline electrolyte increases the risk of corrosion
- Anode: Morphology and bioanodes
- Cathode : Homogeneous catalysis of hydrogen evolution reaction

Amalgamation of Research Areas

- Chemical Engineering
- Electrochemistry
- Sustainable Energy Systems
- Advanced Electronics
- Microbiology
- For Biological systems
 - Process Modelling
 - Machine Learning





THANK YOU FOR YOUR ATTENTION

QUESTIONS